

Revised petrology and new chronological data on the peralkaline felsic lavas of Ngaoundéré volcanism (Adamawa plateau, Cameroon, Central Africa): evidence of open-system magmatic processes

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Abstract

The petrology of the peralkaline felsic lavas from the Ngaoundéré region was re-examined, together with new chronological data obtained on them, in order to determine their petrogenesis and investigate their peralkaline features. The results showed that all lavas have a microlitic porphyritic texture. They are composed of more than 70% phenocrysts and microlites of alkali feldspar, with diopside, hedembergite, augite, aegirine-augite for trachytes, and nosean, nepheline, kaersutite, titanite, apatite and oxides set in a matrix of the same minerals, in phonolite. All lavas, Miocene in age, are the result of the fractional crystallization processes undergone by the surrounding basaltic lavas. The titanite crystallization phase is responsible for the concave downward appearance of the REE pattern of peralkaline phonolite, as well as the slight depletion of Nb and Ta. AFC modeling using Pan-African basement rock as contaminant was also intended. The genesis of peralkaline trachytes from less-evolved lavas requires a Ma/Mc ratio of 0.1, and a fractionation process ($F \approx 0.9$) involving only feldspar so as to comply with the $({}^{87}\text{Sr}/{}^{86}\text{Sr})$ 9Ma values of ≈ 0.710 . Additionally, hydrothermal fluid effects are suspected. The peralkalinity of the lavas is the result of a pronounced fractionation of the Al_2O_3 component by aluminous-enriched phases, in combination with alkali- and volatile-enriched metasomatized fluids, which have acted through the network of Pan-African cracks. Copyright © 2012 Published by Ed. Univ. "Al. I. Cuza" Iași. All rights reserved.

Keywords: petrology, geochronology, peralkaline felsic lavas, crustal contamination, Adamawa-Cameroon, Central Africa.

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Introduction

More than thirty plugs and domes of trachyte, phonolite and peralkaline trachyte and phonolite, surrounded by numerous basaltic lava flows, occur in the north of the Ngaoundéré region, in the Adamawa plateau (Cameroon, Central Africa, Fig. 1). Petrological, geochemical (Nkouandou et al., 2008) and field work data suggest that the felsic lavas are the final products of the differentiation of basaltic lavas through fractional crystallization processes, in spite of the compositional gap between basaltic and felsic lavas. The occurrence of such a gap in alkali volcanism suites has been largely discussed (Thompson et al., 2001) and arose the problem of the origin of the associated felsic lavas. The petrological data on the peralkaline felsic lavas of the Adamawa plateau is not sufficiently detailed. On the other hand, the origin of the peralkaline feature of these lavas is still debated. The following hypotheses have been advanced in this respect: peralkalinity in igneous rocks is attributed to (1) subsolidus reactions with metasomatic fluids (Taylor et al., 1980) or (2) an alkali-enriched source region (Bailey, 1987). It is also suggested that (3) the peralkalinity of felsic magmas may result from the partial melting of different lithospheric mantle or lower continental crustal sources (Bailey and McDonald, 1987). Avanzinelli et (2004) have suggested that al. simple fractional crystallization by itself cannot explain the generation of peralkaline magmas when starting from parental basalts, and that other processes, such as partial melting of gabbroic cumulates, are more viable and have to be considered when it comes to the genesis of peralkaline magmas.



Fig. 1 Geological map of Ngaoundéré region after Nkouandou et al. (2008), modified. Rigth inset shows the mains volcanic zones of Cameroon: Cameroon Volcanic Line and Adamawa plateau.

As a result, the detailed petrology of the peralkaline felsic lavas from the Ngaoundéré region provides an opportunity of highlighting their peralkaline character and the magmatic processes involved in their petrogenesis, using mineral and whole-rock chemistry, as well as Sr-Nd-Pb isotopes. The new chronological data are integrated so as to document the petrogenetic history of the peralkaline felsic lavas from the Adamawa plateau.

Geological setting

The Adamawa Plateau (Fig. 1) is one of the African uplift domes, bordered in the north and in the south by large N70°E striking-slip faults (the Adamawa and Djerem-Mbéré faults respectively), active since the Cretaceous (Moreau et al., 1987). The Adamawa Plateau basement rocks are composed of granitoids facies of Th-U-Pb age - 615 to 575 ± 27 Ma (Tchameni et al., 2006; Ganwa et al., 2008). These rocks are partially covered by Cretaceous-to-recent basaltic lava flows (Lasserre et al., 1961; Temdjim et al., 2004) of typical alkali composition (Nkouandou el al., 2008). The plateau is intruded along the N70E, N65E and N135E Pan-African directions by more than thirty plugs and domes of trachyte and phonolite of peralkaline composition (Fig. 1). Similar alkali lavas series are still studied at the Tchabal Mbabo volcano (Fagny, in preparation), 250 km, western Ngaoundéré. The low partial melting (1 to 2%) of the FOZO-type component mantle source with residual garnet took place at a depth of 80 km (Nkouandou et al., 2010). Numerous basaltic lava flows containing various mantle peridotites (Nkouandou and Temdjim, 2011) are easily distinguished, based on their stratigraphic and surfaceweathering characteristics, as the (1) low flow unit (LFU), (2) middle flow unit (MFU) and (3) upper flow unit (UFU), as established for the basaltic lavas in the nearby Ngaoundéré region (Nono et al., 1994). The felsic lavas sometimes contain fragments (5 to 25 cm in diameter) of basaltic glasses and granitoid basement rock. The origin of these felsic lavas

has been discussed previously (Nkouandou et al., 2008; Legendre et al., 2005). However, it remained difficult to establish, as there is a large compositional gap between the two poles of local volcanism suites. On the other hand, their peralkaline features have never been tackled and they may result from numerous petrogenetic processes, as proven worldwide. The present study deals with the petrology and geochemistry of one peralkaline phonolite plug (Ngao Mokon) and three peralkaline trachyte domes (Ngao Belle, Ngao Madep and Ngao Mboultama), in order to highlight the origin of their peralkaline character and to document their geochronological history.

Analytical methods

Microprobe chemical analyses of the elements (Si, Ti, Al, Cr, Fe, Mn, Mg, Ni, Na, K and P) of representative minerals were carried out using a SX50 and SX100 Camebax microprobe at the "Pierre et Marie Curie" University, Paris 6. The analytical conditions were the following: for clinopyroxene : 15k V and 40 nA, 20s, except Ti (30 s); plagioclase, Kfeldspar, nepheline and noseane: 15k V, 10 nA, 10 s; Fe-Ti oxides: 20 kV and 40 nA: Si. Ca, Ni:10 s; Mn: 25 s ; Cr: 15 s; Al: 30 s; Ti, Fe, Mg: 40 s; kaersutite (analyses not shown): 15 kV, 21 nA; Si, Al, Mg, Na, K, and Ba: 15 s; Ti, Ca: 20 s; Fe, Mn: 25 s; F, Cl: 30 s. The "PAP" correction program used is defined by Pouchou and Pichoir (1991). The analyses of the major elements of all the lavas have been performed through ICP-AES, and those of trace elements - through ICP-MS, at the "Centre de Recherche Pétrographique et Géochimique" of Nancy, France, following the analytical procedure of Carignan et al. (2001). The samples for the isotopic analyses of Rb-Sr and Sm-Nd were dissolved in a mixture of HF-HNO₃ (10:1), and chemical separation was carried out through cationexchange chromatography; blanks were < 1ng. Sr and Nd isotopic ratios were measured on a VG Sector 54 multicollector thermal ionization mass spectrometer. Replicate analyses of the MERCK Nd standard yielded an average $^{143}\text{Nd}/^{144}\text{Nd}$ value of 0.512742 \pm 8 (normalized to $^{146}\text{Nd}/^{144}\text{Nd}$ = 0.7219), and measurements of NBS 987 Sr led to an average $^{87}\text{Sr}/^{86}\text{Sr}$ value of 0.710247 \pm 7 (normalized to $^{86}\text{Sr}/^{88}\text{Sr}$ = 0.1194). Epsilon Nd

values were calculated assuming 147 Sm/ 144 Nd = 0.1967 and 143 Nd/ 144 Nd = 0.512638 for CHUR (after Ashwal et al., 2002). The differentiation index (D.I., Thornton and Tuttle, 1960) is deduced from the CIPW norms, which were calculated assuming a FeO/Fe₂O₃ ratio of 0.3.

Table 1 New chronological data of the Mio-Pliocene alkali lavas series from Ngaoundéré (Adamawa Plateau, Cameroon)

Sample	Rock type	K ₂ O (wt %)	100×rad. ⁴⁰ Ar/total ⁴⁰ Ar	$rad.^{40}Ar$ (10 ⁻¹¹ mol g ⁻¹)	Age (±2Ò) (Ma)
NG130	basalt	2.51	7.6	2.82	7.8 ± 1.4
NG1T	phonolite	5.18	53.1	8.13	10.9 ± 0.4
NG9	peralkaline phonolite	5.38	57.9	7.55	9.7 ± 0.4
NG13	trachyte	5.40	28.1	6.96	8.9 ± 0.4
NG20	peralkaline trachyte	4.80	27.7	5.75	8.3 ± 0.5

Results

Geochronology

The results of the K-Ar dating of the volcanic rocks of the northern and eastern regions of Ngaoundéré are presented in Table 1. All lavas are of Pliocene age, between 7.8 ± 1.4 and 10.9 ± 0.4 Ma, thus within the range of previous K-Ar data (7.0 to 9.8 ± 0.2 Ma; Gouhier et al., 1974) and 40 Ar- 39 Ar data (9.28) to 11.39 ± 0.03 Ma; Marzoli et al., 1999). Among the dated lavas, the felsic are older than the basaltic ones. This proves that the basaltic lavas sampled around the felsic domes cannot be regarded as their parental magma, as they are more recent in age. These observations suggest that all the lavas are from the Miocene eruption period, confirming the occurrence of numerous volcanic episodes.

Field work

One plug of peralkaline phonolite occurs in the northern part of the Ngaoundéré region, while four domes of peralkaline trachyte occur in the eastern part of the region (Fig. 1). The plug and domes occasionally display a strong parallel orientation to the N70E Pan-African direction. They range between 40 to 200 m in diameter, and they crop out between 20 and 250 m above the surrounding basement rocks. A systematic sampling of the lavas has been carried out through painstaking field work. The relative chronology of different rocks is assessed and mostly based upon the xenolith (1 to 20 cm) occurrence of metamorphic, granitic basement rocks or basaltic lava in all lavas. The contact points between plug or domes and granitoid basement rocks are obscured by talus. The plug and domes are always divided into subvertical columns of 1 to 3 meters in diameter, and cut horizontally into units of 2 to 5 meters high. Some display fountains or onion-bulb-like structures with peculiar surfaces on which scattered translucid feldspar phenocrysts (0.5 to 1.0 cm) stand out from the groundmass. The hand specimens display a matrix of greenish or greyish-white colour in trachytic lavas, while phonolite is light green, sometimes bluish when dominated by glassy materials.

Lavas have been named according to their mineralogy (the occurrence of minerals belonging to the nepheline or sodalite groups within phonolite) and their differentiation index (D.I.) after Thornton and Tutle (1960): 93.1 for phonolite, 87.9, 90.1, and 90.7 for peralkaline NG18, NG25 and NG20, respectively. The peralkaline index [P.I. = $(Na_2O + K_2O)/Al_2O_3$] of the lavas is around 1.0 (Table 5), and all the lavas are nepheline normative. SI (diorite) was sampled from well-exposed basement rock of dioritic composition, analysed and added to the present study so as to easily assess the contamination process.

Petrography

Peralkaline phonolites have a microlitic porphyritic texture. They are characterized by the occurrence of abundant needlelike green aegirine crystals, always lying around the phenocrysts and microphenocrysts of minerals belonging to the nepheline and haüyne groups. They also contain abundant titanite (15 vol.%) microphenocrysts in a glassy matrix of microcrystalline aegirine, alkali feldspar, rare haüyne group minerals, nepheline and interstitial oxides, and apatite. Nepheline microphenocrysts occur as euhedral hexagonal crystals commonly containing acicular deep-green aegirine. Amphibole phenocrysts of the kaersutite type (analyses not shown), rounded by a thin layer of Fe-Ti oxides, sometimes contain apatite or titanite. Alkali feldspar phenocrysts dominate the mineral assemblage. Two types of haüyne group mineral phenocrysts, up to 2 mm in size occur: thin light rime with a dark-greyish core, and dark rime (containing numerous pyrite microcrysts) with a white-greyish core. Phenocrysts always form a glomeroporphyritic texture within the glassy matrix. Zeolite veins (1–2 mm wide) are common.

Peralkaline trachytes exhibit a microlitic porphyritic texture, characterized by the occurrence of abundant (> 70 wt%) K-feldspar phenocrysts and microphenocrysts, set in the groundmass made up of the same phases, plus Fe-Ti oxides. K-feldspar crystals are rounded by numerous fibrous deep-green aegirine microlites. Both euhedral diopside and rare dark-green skeleton aegirine-augite phenocrysts (0.5 mm) occur within the interstitial matrix of K-feldspar phenocrysts. Titanite, Fe-Ti oxides and apatite phenocrysts are also present in some lavas. Microlites of the same minerals constitute the matrix of all the lavas, except peralkaline trachyte NG20, which is only composed of K-feldspar, aegirine-augite and rare Fe-Ti oxides (< 1 wt%).



Fig. 2 Evolutionary trends of clinopyroxene compositions of peralkaline felsic lavas from Ngaoundéré plotted in terms of : (a) Na–Mg– (Mn + Fe²⁺) and (b) Wo–En–FS.

Mineralogy

The clinopyroxene analyses conducted on all the lavas are presented in Table 2 and

represented in Figs. 2a and 2b. Their compositions (nomenclature after Morimoto et al., 1988) vary (Fig. 2a) from diopside (Wo_{46.0-48.2}En_{34.8-36.1}) to hedenbergite ($Wo_{46.3-47.7}Fs_{23.0-47.7}$) in peralkaline trachyte, while augite ($Wo_{39.7-32.0}En_{26.1-10.9}$), aegirine-augite ($Wo_{33.1-39.7}Fs_{66.4-34.2}$) and aegirine ($Wo_{29.4}Fs_{66.4}$) occur in peralkaline phonolite and trachyte. All compositions define a regular evolutionary trend, as shown in Fig. 2b. Aegirine-augite in peralkaline phonolite are rich in TiO₂ (up to 6.1 wt% TiO₂ in microlites), while high contents of Fe₂O₃* (up to 33.0 wt%) and Na₂O (12.3 wt%) are found in peralkaline trachyte microlites with a high value of acmite (Ac₈₃). Prior to interpretation, high TiO₂ contents (6.1 wt%) in aegirine microlites (with Al_{VI}/Al_{IV} < 0.25) indicate a low crystallization temperature (< 600°C; Ferguson, 1977), and the occurrence of Fe–NAT (NaTi_{0.5}Fe²⁺_{0.5}Si₂O₆) in microlites suggests low oxygen fugacity during crystallization (Nielsen, 1979; Duggan, 1988).

Table 2 Representative chemical analyses of clinopyroxene of peralkaline felsic lavas from Ngaoundéré; ph, c: phenocryst core; ph, r: phenocryst rime; ml: microlite

Lava	Peralk. t	rachyte							Peralk. pl	honolite	
Sample	NG20		NG25		NG18				NG9		
Туре	ph.c			ph.r	ph.c			ml	ph.c	ml	
	Oxides (wt%)										
SiO ₂	52.35	53.23	53.21	51.60	51.62	52.37	51.35	53.54	53.33	53.24	
TiO ₂	0.37	1.82	1.39	0.55	0.53	0.49	0.88	6.07	5.13	4.79	
Al_2O_3	0.64	1.14	0.75	2.20	2.22	0.70	1.72	0.79	0.55	0.83	
FeO	33.12	31.20	32.04	12.54	12.61	13.52	13.05	21.90	24.39	25.24	
MnO	0.89	0.95	1.14	1.88	1.63	2.07	2.21	0.96	1.16	1.33	
MgO	0.11	0.10	0.10	9.47	9.56	8.94	8.91	3.07	1.67	1.40	
CaO	2.55	2.23	2.94	19.43	19.41	20.12	18.88	4.91	3.26	2.64	
Na ₂ O	12.19	12.33	12.04	1.98	2.03	1.63	2.80	11.25	11.62	12.32	
ZrO ₂	0.39		0.06								
Sum	102.61	102.99	103.68	99.65	99.61	99.84	99.80	102.49	101.11	101.79	
	Cations (apuf)										
Si	1.976	1.989	1.984	1.962	1.962	2.004	1.950	1.975	2.008	1.989	
Ti	0.010	0.051	0.039	0.016	0.015	0.014	0.025	0.168	0.145	0.135	
$Al_{\rm VI}$	0.005	0.040	0.017	0.061	0.061	0.032	0.027	0.034	0.024	0.037	
Al _{IV}	0.024	0.011	0.016	0.038	0.038		0.050				
Fe ³⁺ _{VI}	0.882	0.762	0.789	0.092	0.097	0.053	0.176	0.481	0.516	0.608	
Fe ²⁺	0.065	0.129	0.123	0.297	0.293	0.374	0.219	0.141	0.194	0.113	
Mn	0.029	0.030	0.036	0.061	0.052	0.067	0.071	0.030	0.037	0.042	
Mg	0.006	0.006	0.006	0.537	0.542	0.510	0.504	0.169	0.094	0.078	
Ca	0.103	0.089	0.118	0.792	0.790	0.825	0.768	0.194	0.131	0.106	
Na	0.892	0.894	0.871	0.146	0.150	0.121	0.206	0.805	0.848	0.893	
Zr	0.007		0.001								
				Μ	linerals (%)					
Wo	47.7	33.1	39.8	46.0	46.1	46.3	48.2	39.7	29.4	32.0	
En	4.6	2.4	0.3	34.8	35.0	30.7	36.1	26.1	4.2	10.9	
Fs	47.7	64.5	59.9	19.2	18.9	23.0	15.7	34.2	66.4	57.1	

Lavas	Peralk	aline trac	hyte							Peralk	aline phor	nolite					
Samples	NG25				NG20		NG18		-	NG9							
Minerals	K-feld	spar								Nephe	eline		Nose	eane			
Description	ph.c								-	ph.c.	mph	-		d.c.	l.r.	l.c.	d.r.
SiO ₂ (wt%)	67.03	66.56	67.23	67.38	69.00	69.32	68.73	69.16	68.94	46.96	47.65	49.51	36.84	36.36	37.13	36.71	33.37
Al_2O_3	18.97	19.04	19.23	18.97	19.35	19.56	19.69	19.77	19.47	32.53	31.51	29.83	30.22	29.97	28.79	27.95	26.97
FeO*	0.30	0.18	0.24	0.14	0.38	0.31	0.18	0.32	0.34	0.84	0.70	1.88	0.96	0.25	0.23	2.16	4.50
CaO	0.15	0.13	0.18	0.18	0.07	0.15	0.10	0.12	0.09	0.25	0.28	0.08	0.48	0.14	0.98	0.68	0.36
Na ₂ O	7.01	5.65	6.71	7.39	8.49	8.79	9.01	8.91	8.95	16.84	16.24	16.54	18.17	21.33	20.01	16.72	14.02
K ₂ O	6.43	8.27	6.56	6.04	5.49	4.61	4.85	4.57	4.87	3.56	3.58	2.96	2.01	1.86	3.51	7.93	11.24
SO ₃													10.11	9.76	8.20	7.27	9.71
Cl													0.57	0.79	0.80	0.35	0.07
Sum	99.89	99.83	100.15	100.1	102.78	102.74	102.56	102.85	102.66	100.98	99.95	100.80	99.36	100.46	99.65	99.77	100.24
Si (apuf)	2.996	2.993	2.995	3.001	2.990	2.993	2.980	2.982	2.986	8.773	8.781	9.182	6.040	6.070	6.260	6.180	5.840
Al	0.999	1.009	1.010	0.996	0.988	0.995	1.006	1.005	0.994	7.163	7.145	6.521	5.840	5.900	5.720	5.550	5.570
Fe ⁺⁺	0.011	0.007	0.009	0.005	0.014	0.011	0.007	0.011	0.012	0.126	0.107	0.265	0.120	0.030	0.030	0.270	0.590
Ca	0.007	0.006	0.009	0.009	0.003	0.007	0.005	0.006	0.004	0.050	0.057	0.016	0.080	0.020	0.180	0.120	0.070
Na	0.609	0.494	0.581	0.640	0.715	0.738	0.760	0.747	0.754	6.118	6.075	5.964	5.780	6.910	6.530	5.460	4.760
К	0.367	0.474	0.373	0.343	0.304	0.254	0.268	0.251	0.269	0.847	0.878	0.701	0.420	0.400	0.750	1.700	2.510
S													1.240	1.220	1.040	0.920	1.280
Cl													0.160	0.230	0.230	0.100	0.020
Or (%)	37.3	48.7	38.7	34.4	29.7	25.4	25.9	25	26.2								
Ab	62	50.7	60.4	64.2	70	73.8	73.5	74.4	73.3								
An	0.7	0.6	0.9	1.4	0.3	0.8	0.6	0.6	0.5								
Ne										79.1	78.9	66.7					
Ks										14.8	15	22.3					
Q										6.1	6.1	11					

Table 3 Representative chemical analyses on feldspar and feldspatoids from the peralkaline felsic lavas of Ngaoundéré; l.r.: light rime; l.c.: light core; d.r.: dark rime; d.c.: dark core; ph.c: phenocryst core; mph: microphenocryst

Minerals Titanite Titanignetite Samples NG9 NG2 NG18 NG25 NG2 NG10 29.44 29.75 29.58 29.86 1.4.29 14.53 StO2 30.00 29.44 29.75 29.58 29.86 14.53	Lavas	Peralk phonolite	Peralk. trachyte						Pera	ılk. olite	
Samples NG9 NG29 Ph.c NG18 NG25 S13.6 33.26 33.26 33.26 33.22 12.83 13.16 16.38 14.29 14.53 ZrO2 0.69 0.82 0.76 1.14 0.68 1.12 1.58 1.83 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.75 1.78 0.19 0.60 0.13 1.02 0.31 Y2O3 0.62 0.73 0.71 0.48 0.61 0.43 3.38 3.41 3.35 Al2O3 1.65 1.87 1.58 1.75 1.78 0.02 0.13	Minerals	Titanite	2				Ti-mag	netite			
Descriptionph.cmicph.cph.cph.cph.c(wt%)(wt%)SiO230.0029.4429.5829.86TiO233.5733.2633.2212.8313.1616.3814.29JAC0.690.820.761.140.681.15Nb_0O51.121.831.811.811.810.690.610.471.140.680.610.471.140.620.730.710.480.610.47V2000.610.4131.801.900.620.730.710.430.433.383.413.600.620.610.610.610.610.610.730.730.730.730.730.730.730.730.730.730.73 <td< td=""><td>Samples</td><td>NG9</td><td>NG20</td><td></td><td>NG18</td><td>NG25</td><td>NG18</td><td>NG25</td><td>NG9</td><td></td><td></td></td<>	Samples	NG9	NG20		NG18	NG25	NG18	NG25	NG9		
(wt%)SiO2 30.00 29.44 29.75 29.58 29.86 TiO2 33.57 33.26 33.26 33.22 13.83 $1.31.6$ 16.38 14.29 14.53 ZrO2 0.69 0.82 0.76 1.14 0.68 1.28 13.16 16.38 14.29 14.53 Nb05 1.12 1.58 1.83 1.81 1.75 1.58 1.58 1.58 1.58 1.58 1.58 1.58 1.58 1.57 1.58 1.57 1.58 1.57 1.78 0.19 0.60 0.13 1.02 0.31 Nd2O3 0.62 0.73 0.71 0.48 0.61 0.50 0.13 1.02 0.31 Y2O3 0.24 0.39 0.36 0.22 0.29 0.60 0.13 1.02 0.31 Fe ₅ O3 2.42 2.59 2.39 2.50 2.43 81.30 80.91 78.89 78.18 79.96 MnO T 3.69 4.33 3.38 3.41 3.35 CaO 26.42 25.95 25.85 25.82 25.67 98.03 99.14 96.78 96.93 98.19 MagO 0.060 0.07 0.09 0.07 T T $S1.64$ 3.49 3.428 3.497 0.161 0.165 0.205 0.179 0.182 Zr 0.047 0.056 0.058 0.058 0.058 0.055 1.57 5.57	Description	ph.c		ph.c		mic	ph.c		ph.c		
SiO230.0029.4429.7529.5829.86TiO233.5733.2633.2633.2212.8313.1616.3814.2914.53ZrO30.690.820.761.140.68 $\cdot \cdot $					(wt%	6)					
TiO233.5733.2633.2633.2633.2212.8313.1616.3814.2914.53ZrO20.690.820.761.140.68	SiO ₂	30.00	29.44	29.75	29.58	29.86					
ZrO20.690.820.761.140.68Nb2051.121.581.831.811.75La2O30.370.510.600.610.47Ce2O30.690.880.830.790.83Nd2O30.620.730.710.480.61Y2O30.240.390.360.220.29Fe;O31.651.781.751.780.190.600.131.020.31Fe;O32.422.592.392.502.4381.3080.9176.8978.1879.69MnO	TiO ₂	33.57	33.26	33.26	32.26	33.22	12.83	13.16	16.38	14.29	14.53
Nb ₂ O51.121.581.831.811.75La ₂ O ₃ 0.370.510.600.610.47Ce ₂ O ₃ 0.690.880.790.83Nd ₂ O ₅ 0.620.730.710.480.61Y ₂ O ₃ 0.240.390.360.220.29Al ₂ O ₃ 1.651.871.581.751.780.190.600.131.020.31Fe ₂ O ₃ 2.642.592.592.582.6381.0380.9176.8978.1879.09MnO	ZrO ₂	0.69	0.82	0.76	1.14	0.68					
La2O30.370.510.600.610.47Ce2O30.690.880.830.790.83Nd2O30.620.730.710.480.61Y2O30.240.390.360.220.29Al2O31.651.871.581.751.780.190.600.131.020.31Fe2O32.422.592.392.502.4381.3080.9176.8978.1879.66MnO3.694.333.383.413.353.633.030.35CaO2.6422.5952.58525.8225.670.020.150.000.030.03Na2O0.060.070.090.07 </td <td>Nb₂O5</td> <td>1.12</td> <td>1.58</td> <td>1.83</td> <td>1.81</td> <td>1.75</td> <td></td> <td></td> <td></td> <td></td> <td></td>	Nb ₂ O5	1.12	1.58	1.83	1.81	1.75					
Ce_2O_3 0.690.880.830.790.83 Nd_2O_3 0.620.730.710.480.61 Y_2O_3 0.240.390.360.220.29 Al_2O_3 1.651.871.581.751.780.190.600.131.020.31 Fe_2O_3 2.422.592.392.502.4381.3080.9176.8978.1879.96 MnO	La ₂ O ₃	0.37	0.51	0.60	0.61	0.47					
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	Ce ₂ O ₃	0.69	0.88	0.83	0.79	0.83					
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Nd ₂ O ₃	0.62	0.73	0.71	0.48	0.61					
Al2O31.651.871.581.751.780.190.600.131.020.31Fe2O32.422.592.392.502.4381.3080.9176.8978.1879.96MnO	Y ₂ O ₃	0.24	0.39	0.36	0.22	0.29					
Fe_2O_3 2.422.592.392.502.4381.3080.9176.8978.1879.96MnO3.694.333.383.413.35CaO26.4225.9525.8525.8225.670.020.150.000.030.03Na2O0.060.070.090.07Sum97.8598.0898.0297.0397.5798.0399.1496.7896.9398.19Si4.1684.1244.1624.1794.180Si4.1684.1244.1624.1794.180Ti3.5073.5043.4993.4283.4970.1610.1650.2050.1790.182Zr0.0470.0560.0520.0790.046Y0.0090.0150.0130.0080.011Al0.0150.0130.0160.012Mb0.0350.0500.0580.0580.0510.0130.0010.011Al0.0130.0130.0160.012	Al ₂ O ₃	1.65	1.87	1.58	1.75	1.78	0.19	0.60	0.13	1.02	0.31
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Fe ₂ O ₃	2.42	2.59	2.39	2.50	2.43	81.30	80.91	76.89	78.18	79.96
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	MnO						3.69	4.33	3.38	3.41	3.35
Na_2O 0.060.070.090.07Sum97.8598.0898.0297.0397.5798.0399.1496.7896.9398.19(aptiv)Si4.1684.1244.1624.1794.180 $V = V = V = V = V = V = V = V = V = V =$	CaO	26.42	25.95	25.85	25.82	25.67	0.02	0.15	0.00	0.03	0.03
Sum97.8598.0898.0297.0397.5798.0399.1496.7896.9398.19Si4.1684.1244.1624.1794.180 \times <td< td=""><td>Na₂O</td><td>0.06</td><td>0.07</td><td>0.09</td><td>0.07</td><td></td><td></td><td></td><td></td><td></td><td></td></td<>	Na ₂ O	0.06	0.07	0.09	0.07						
tabularSi4.1684.1244.1624.1794.180Ti3.5073.5043.4993.4283.4970.1610.1650.2050.1790.182Zr0.0470.0560.0520.0790.046 $$	Sum	97.85	98.08	98.02	97.03	97.57	98.03	99.14	96.78	96.93	98.19
Si4.1684.1244.1624.1794.180Ti 3.507 3.504 3.499 3.428 3.497 0.161 0.165 0.205 0.179 0.182 Zr 0.047 0.056 0.052 0.079 0.046 $$					(apf	u)					
Ti 3.507 3.504 3.499 3.428 3.497 0.161 0.165 0.205 0.179 0.182 Zr 0.047 0.056 0.052 0.079 0.046 $$	Si	4.168	4.124	4.162	4.179	4.180					
Zr0.0470.0560.0520.0790.046Nb0.0350.0500.0580.0580.055	Ti	3.507	3.504	3.499	3.428	3.497	0.161	0.165	0.205	0.179	0.182
Nb0.0350.0500.0580.0580.055Y0.0090.0150.0130.0080.011La0.0090.0130.0150.0160.012Ce0.0180.0230.0210.021Nd0.0150.0180.0120.015Al0.1350.1540.1300.1460.1470.0040.0120.003 Fe^{3+} 0.2530.2730.2520.2661.0701.0651.0191.0371.056Mn0.2520.2660.2561.0700.0440.0130.014Mg3.8953.8743.9093.850 </td <td>Zr</td> <td>0.047</td> <td>0.056</td> <td>0.052</td> <td>0.079</td> <td>0.046</td> <td></td> <td></td> <td></td> <td></td> <td></td>	Zr	0.047	0.056	0.052	0.079	0.046					
Y 0.009 0.015 0.013 0.008 0.011 La 0.009 0.013 0.015 0.016 0.012 0.012 Ce 0.018 0.023 0.021 0.021 0.021 Nd 0.015 0.018 0.012 0.015 0.015 0.012 Al 0.135 0.154 0.130 0.146 0.147 0.004 0.012 0.003 0.020 0.006 Fe ³⁺ 0.253 0.273 0.252 0.266 0.256 1.070 1.065 1.019 1.037 1.056 Mn $$	Nb	0.035	0.050	0.058	0.058	0.055					
La0.0090.0130.0150.0160.012Ce0.0180.0230.0210.0200.021Nd0.0150.0180.0120.0150.015Al0.1350.1540.1300.1460.1470.0040.0120.0030.0200.006 Fe^{3+} 0.2530.2730.2520.2660.2561.0701.0651.0191.0371.056Mn- 0.0520.0610.0480.0480.0480.047Mg- 0.0010.0040.0000.0010.001Ca3.9323.8953.8743.9093.850	Y	0.009	0.015	0.013	0.008	0.011					
Ce 0.018 0.023 0.021 0.020 0.021 Nd 0.015 0.018 0.012 0.015 0.015 Al 0.135 0.154 0.130 0.146 0.147 0.004 0.012 0.003 0.020 0.006 Fe ³⁺ 0.253 0.273 0.252 0.266 0.256 1.070 1.065 1.019 1.037 1.056 Mn 0.025 0.266 0.256 0.061 0.048 0.048 0.047 Mg 3.874 3.909 3.850 </td <td>La</td> <td>0.009</td> <td>0.013</td> <td>0.015</td> <td>0.016</td> <td>0.012</td> <td></td> <td></td> <td></td> <td></td> <td></td>	La	0.009	0.013	0.015	0.016	0.012					
Nd 0.015 0.018 0.018 0.012 0.015 Al 0.135 0.154 0.130 0.146 0.147 0.004 0.012 0.003 0.020 0.006 Fe ³⁺ 0.253 0.273 0.252 0.266 0.256 1.070 1.065 1.019 1.037 1.056 Mn 0.012 0.004 0.012 0.003 0.020 0.006 Mg 0.052 0.061 0.048 0.048 0.047 Mg 3.895 3.874 3.909 3.850 </td <td>Ce</td> <td>0.018</td> <td>0.023</td> <td>0.021</td> <td>0.020</td> <td>0.021</td> <td></td> <td></td> <td></td> <td></td> <td></td>	Ce	0.018	0.023	0.021	0.020	0.021					
	Nd	0.015	0.018	0.018	0.012	0.015					
Fe ³⁺ 0.253 0.273 0.252 0.266 0.256 1.070 1.065 1.019 1.037 1.056 Mn	Al	0.135	0.154	0.130	0.146	0.147	0.004	0.012	0.003	0.020	0.006
Mn 0.052 0.061 0.048 0.048 0.047 Mg 0.001 0.004 0.000 0.001 0.001 0.001 Ca 3.932 3.895 3.874 3.909 3.850	Fe ³⁺	0.253	0.273	0.252	0.266	0.256	1.070	1.065	1.019	1.037	1.056
Mg 0.001 0.004 0.000 0.001 0.001 Ca 3.932 3.895 3.874 3.909 3.850 - <td>Mn</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>0.052</td> <td>0.061</td> <td>0.048</td> <td>0.048</td> <td>0.047</td>	Mn						0.052	0.061	0.048	0.048	0.047
Ca 3.932 3.895 3.874 3.909 3.850 Na 0.008 0.009 0.013 0.009 X'Usp 0.36 0.36 0.47 0.44 0.41	Mg						0.001	0.004	0.000	0.001	0.001
Na 0.008 0.009 0.013 0.009 X'Usp 0.36 0.36 0.47 0.44 0.41	Ca	3.932	3.895	3.874	3.909	3.850					
X'Usp 0.36 0.36 0.47 0.44 0.41	Na	0.008	0.009	0.013	0.009						
	X'Usp						0.36	0.36	0.47	0.44	0.41

Table 4 Representative chemical composition of titanite, Ti-magnetite and apatite of Ngaoundéré peralkaline felsic lavas; ph.c.: phenocryst core; mic: microcryst

The analyses conducted on representative alkali feldspar in all the lavas reveal sanidine $(Or_{37,3-48.7}Ab_{62.0-50.7})$ and anorthose $(Or_{25.0-34.4}Ab_{74.4-64.2})$ in composition, as shown in

Figure 3 and presented in Table 3. Albite phenocrysts (analyses not shown) also occur in some peralkaline phonolites. The Al_2O_3 content is around 20 wt% in all the feldspar

crystals analysed. The mean of $Na_2O + K_2O$ content is 13.0 wt%.



Fig. 3 Feldspar chemical composition ploted in Or–Ab–An diagram of peralkaline felsic Miocene lavas from Ngaoundéré.

The nepheline composition (Table 3) ranges between Ne_{66.7–79.1} Ks_{22.3–14.8} Qz_{6.1–11.0}. The average Na₂O content is 16.5 wt%, while SiO₂, K₂O and recalculated Fe₂O₃ contents are high (up to 50.0, 3.6 and 1.9 wt%, respec-tively). The Al₂O₃ content is up to 33 wt%. The crystallization temperature of nepheline is higher than 700°C (estimated in agreement with Hamilton, 1961). Petrographic observa-tions indicate that nepheline crystallized as a primary phase.

The minerals of the Haüyne group are markedly heterogeneously nosean in composition, with SO₃ contents between 7.0 and 10.1wt %, the lowest value being recorded in light cores and rims (Table 3). The Al₂O₃ and Na₂O + K₂O contents range between 27.0–30.2 wt% and 20.2–25.2 wt%, respectively. In anticipation, more phenocrysts are euhedral, indicating their crystallization as primary phase (Wooley and Platt, 1988). Dark-grey crystals with a light rim are distinguished from light-grey ones with a dark-grey rim. Dark rims contain numerous pyrite microcrystals.

In advance, the noseane + nepheline + sanidine association indicates low crystallization pressures (0.16 < P GPa < 0.28, estimated according with Wellman, 1970). Titanite (Table 4) phenocrysts, microphenocrysts and microlites occur in all the lavas (TiO₂: 32.0-33.3, and CaO: 25.0-26.4 wt %), and are sometimes as abundant as 10 % of the total volume in peralkaline phonolite. Zr (4928.6–8143.0 ppm) and Nb (4223.0-6824.6ppm) contents are high at Ngao Makan (NG9). Zr/Nb ratios are lower than 2.0.

Ti-magnetite (TiO₂: 12.0-16.4 wt%, recalculated Fe₂O₃ between 78.0 and 81.3 wt%; Table 4) is present as microcrysts in all the lavas. MnO contents are up to 4.3 wt%.

Apatite is present in peralkaline phonolite as fluoroapatite (CaO: 53.45-53.50 wt%, and P₂O₅: 39.44-39.74 wt%), with F contents of up to 3.5 wt%.

Geochemistry

Whole-rock chemistry and isotopic analyses of all the lavas with surrounding dioritic basement rock are presented in Table 5. Four main oxides constitute more than 90 wt% of the major elements: SiO₂ (56.0–62.6 wt%), Al₂O₃ (16.8–22.4 wt%), with (Na₂O + K₂O) between 12.7 and 15.7 wt% (the higher value being recorded in peralkaline phonolite). Apart from Fe₂O₃, which is up to 4.4 wt%, early fractionated oxides remain low (< 2.0 wt%).

Transition element contents (Ni, Cr, V and Co) are usually below the analytical detection limit. Rb and Cs are very high in peralkaline trachytes (Rb = 350 ppm; Cs = 4.55 ppm). Sr and Ba contents are low in all the lavas, while contents of Be and incompatible elements (Th, U, Nb, Ta, Pb, Hf, Zr) are high in peralkaline trachytes. The highest Zr (2004 ppm) and Nb (725 ppm) contents are found in peralkaline trachytes. Zr/Hf ratios vary widely between 38.3 and 57.4, while the Nb/Ta ratio ranges between 13.7 and 25.1, as indicated by their patterns.

As a typical feature of the alkaline lavas series worldwide, REE normalized patterns (Fig. 4a, normalization values after McDonough and Sun, 1995) exhibit a regular decrease from La to Lu, with significant enrichment of LREE (100 to 300 times the chondritic normalized values) and tend towards flat heavy REE. Lavas are characterized by concave-downward REE patterns, particularly for peralkaline phonolite NG9, which is also characterized by the lack of elementary anomalies and lower REE abundances, as depicted in Figure 4a. In addition, this lava is characterized by an increasing depletion in MREE. Peralkaline trachytes (particularly NG20) have high REE contents and a noteworthy Eu anomaly (Eu/Eu* as low as 0.6). (La/Yb)_N ratios vary between 13.6 and 22.5.

Table 5 whole fock geochemistry of perarkine feisie lavas and basement fock from regiounder	Table 5 Whole rock geochemi	stry of peralklin	e felsic lavas and basem	ent rock from Ngaoundére
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Lavas	Peralkaline phonolite	Peralkaline trachyte			Diorite
Samples	NG9	NG18	NG20	NG25	SI
		Oxides (wt%)			
SiO ₂	56.01	60.63	62.58	61.70	53.47
TiO ₂	0.30	0.29	0.19	0.23	1.01
Al_2O_3	22.43	18.19	16.75	18.07	17.87
Fe ₂ O ₃	2.13	3.86	4.43	3.74	7.98
MnO	0.18	0.42	0.57	0.37	0.13
MgO	0.09	0.22	0.03	0.18	3.83
CaO	0.72	1.17	1.02	0.92	7.09
Na ₂ O	10.16	7.38	7.71	8.04	4.17
K ₂ O	5.66	5.25	4.76	4.95	2.15
P_2O_5	0.08	0.05	0.05	0.05	0.35
LOI	1.39	1.62	1.48	1.28	1.22
Sum	99.14	99.08	99.57	99.53	99.25
D.I.	93.1	87.9	90.1	90.7	48.0
P. I.	1.0	1.0	1.1	1.0	0.5
		Cations (ppm)			
Ве	7.0	9.3	19.0	11.6	2.9
Rb	240	210	350	240	64
Sr	201	45	31	28	755
Cs	2.25	1.57	4.45	1.51	2.19
Ва	140	133	15	20	747
V	11				161
Cr	< L.D.				41.6
Co	0.4	0.3	0.2	0.2	21.7
Ni	< L.D.				22.8
Cu	< L.D.				17
Zn	141	245	406	253	97
Y	17.6	75.1	137.5	69.7	20.3
Zr	1,244	1,307	2,004	1,358	175
Nb	172	347	725	419	10
Hf	21.7	32.1	52.4	33.9	4.4
Та	6.88	25.16	50.51	30.47	0.59
Th	28.19	34.66	85.86	42.09	1.77
U	11.57	9.27	21.29	11.50	0.83

Lavas	Peralkaline phonolite	Peralkaline trachyte			Diorite
Samples	NG9	NG18	NG20	NG25	SI
La	57.4	239.7	249.8	201.2	27.7
Ce	88	426	524	371	61
Pr	7.7	39.6	53.5	34.6	8.0
Nd	22	133	189	115	33
Sm	3.3	19.8	32.1	17.7	6.5
Eu	0.94	3.01	3.38	2.19	1.71
Gd	2.63	14.20	23.49	13.03	4.99
Tb	0.436	2.239	3.851	2.089	0.705
Dy	2.79	12.64	22.37	12.01	3.86
Но	0.58	2.44	4.29	2.29	0.70
Er	1.85	7.02	12.29	6.72	1.92
Tm	0.33	1.07	1.89	1.05	0.28
Yb	2.56	7.24	12.46	7.14	1.82
Lu	0.44	1.10	1.82	1.09	0.29
⁸⁷ Sr/ ⁸⁶ Sr	0.703614	0.706072	0.710526		0.706886
143Nd/144Nd	0.512826	0.512742	0.512754		0.512312
²⁰⁶ Pb ^{/204} Pb	20.2353	19.038	19.2958		17.8948

In spider diagrams (Fig. 4b), patterns are similar to typical OIB, with positive Nb–Ta and negative Ba, K, Sr, and Ti anomalies. They are also characterized by positive anomalies in the case of Zr and Hf. Peralkaline phonolite is marked by slightly negative Nb and Ta anomalies.

Strontium. Nd and Pb isotopic ratios have measured for the selected been lavas (peralkaline phonolite and trachyte) and in the Ngaoundéré basement rock (Tab. 5). Initial isotopic compositions (Sr_i) have been recalculated for 9 Ma. The isotopic ratios of peralkaline phonolite are within the 0.703390 (± 0.00009) range. Peralkaline trachytes NG18 and NG20 have distinct highest values: 0.706072 (± 0.000009) and 0.710526 (± 0.000008), respectively.

The peralkaline felsic lavas of Ngaoundéré exhibit a slight difference in the case of Pb and Nd isotopes, compared to those erupted along the Cameroon Line (Halliday et al., 1988). Peralkaline phonolite with a low ⁸⁷Sr/⁸⁶Sr isotope value has the highest ²⁰⁶Pb/²⁰⁴Pb and ¹⁴³Nd'¹⁴⁴Nd values, compared to peralkaline trachytes NG18 and NG20 (see

Table 3), which are more radiogenic in the ⁸⁷Sr/⁸⁶Sr isotope, close to diorite sample S1 from the neighbouring Pan-African basement.

The comparison of regional variations (Halliday et al., 1990; Lee et al., 1994; Yokoyama et al., 2006) shows that the isotopic compositions of the Ngaoundéré peralkaline felsic lavas display slightly different values. For example, Pb ratios ($^{206}Pb/^{204}Pb$: 19.04–20.24) are intermediate between those from volcanoes located near the continent-ocean boundary (Bioko, Mt. Etinde and Mt. Cameroon: $^{206}Pb/^{204}Pb$ = 20.52) and that of continental volcanoes (e.g. Pagalù, the southernmost island: $^{206}Pb/^{204}Pb$ = 19.01).

Discussions

Petrogenesis and geochronology of the felsic lavas from Ngaoundéré

There is no doubt that the occurrence of basaltic and felsic lavas side by side in the area, as well as the presence of fragments of basaltic glasses in the felsic lavas, are solid field arguments, which clearly suggest a genetic relationship between the lavas.



(a)



Fig. 4 Normalized REE (a) and multi-element patterns (b) to primitive mantle (after McDonough and Sun, 1995) of peralkaline phonolite (NG9), trachytes (NG20, NG25 and NG18) and diorite basement rock (SI).

This plausible genetic relationship has previously been modelled by Nkouandou et al. (2008) when discussing the petrogenesis of the Mio-Pliocene lava series of the Adamawa region. It was, thus, suggested that the felsic lavas derived from basaltic ones through fractional crystallization processes, in spite of the compositional gap between felsic and basaltic lavas. New K-Ar dating of volcanic rocks of the North and East of Ngaoundéré challenge this assumption, as the former are older (8.3 to 11.0 Ma) than the latter (7.8 Ma). Regarding the petrographic study, the modal mineralogical composition is mainly dominated by alkali feldspar (up to more than 70% of the total volume), nosean and nepheline (in phonolite), augite, aegirine-augite and aegirine phases, which are believed to occur in the late differentiated lavas of the alkaline magmatic series, as indicated by the temperature and pressure present during the crystallization of these minerals.

Geochronological data, as well as the existing compositional gap between basaltic and felsic lavas, raise the question of the origin of the peralkaline felsic lavas from Ngaoundéré volcanism. Three main hypotheses may be suggested in this respect: (1) Ngaoundéré peralkaline felsic lavas may represent the primary melt of the local basement rock or the (2) end-product of the fractional crystallization process of the primitive mantle melt, as previously believed. As alternative hypothesis, the peralkaline felsic lavas of the Adamawa volcanic province may be the result of the (3) late combination of the complexing and transport of the volatile fluids phase with fractional crystallization. The petrogenetic implication of crustal anatexis (first proposal) encounters difficulties in explaining the process of generation of peralkaline melts from the crustal lithologies of the Ngaoundéré basement rocks, which are metaluminous or peraluminous (Tchameni et al., 2006; Ganwa et al., 2008); it also lacks experimental verification. In addition, the high $(La/Yb)_N$ values suggest the low-degree partial melting of the mantle source, situated at more than 80 km, as suggested by the low content of heavy REE, indicating the presence of residual garnet in the parental mantle source. Positive Nb and Ta anomalies in the pattern of trace elements point against crustal source melting. Thus, the peralkaline felsic lavas of Ngaoundéré may be regarded as the products of the differentiation process of the surrounding basaltic lavas, generated within the spinel-peridotite stability field of the FOZO type component, at depths below 90 km, in the lithospheric mantle source, as previously suggested bv Nkouandou et al. (2010). The very low contents of early fractionated transional elements (Ni, Co and Cr), the high contents of incompatible elements (Zr, Nb, LREE etc.), together with the continuous chemical variation of the clinopyroxene phases, strongly support this hypothesis. However, the contamination effect of peralkaline trachyte (NG20 and NG18) is relevant, as proven below. From a geochronological point of view, we simply suggest that the analysed basalt has probably been sampled from the basaltic lavas of the upper flow unit, which certainly erupted after the construction of the felsic lava domes.

Taking into consideration the isotopic data available on the lavas, peralkaline phonolite NG9, with a (Sr_i) value of approximately 0.703, is of mantle origin, as this value is isotopically similar to those of the basaltic and felsic lavas from the Cameroon Volcanic Line (Lee et al., 1996; Marzoli et al., 2000; Wandji et al., 2008; Kamgan et al., 2010; Déruelle et al., 2007) and the Adamawa plateau (Nkouandou et al., 2010), which are already regarded as mantle products. Among the Sr, Nd and Pb isotopes, only Sr isotopes suggest a possible contamination of peralkaline trachytes by crustal materials or fluid (Macdonald, 1987; Cousens et al., 1993). Peralkaline trachyte NG18, with a Sr_i value of 0.706, close to that of Ngaoundéré basement rock (represented here by sample SI), might suggest a crustal contamination/assimilation process involving SI as the protolith. Peralkaline trachyte NG20, with a higher Sr_i value (0.710), might have undergone a complex petrogenetic process, with (1) the involvement of other crustal contaminating materials, different in composition

from SI, or even (2) a primary melt of the crustal region with such a Sr_i composition. These suggestions are strongly supported by the occurrence of fragments of country rocks of different compositions (granulites and gneiss), which are potential protoliths for partial melting. An alternative hypothesis for the genesis of peralkaline trachyte NG20 is an addition to crustal contamination materials of a volatile fluids phase, through complexing and transport, as previously suggested by Ren et al. (2006) for the peralkaline trachytes and pantellerites of the Eburru volcanic complex. This suggestion is the potential solution when trying to explain the high Sr_i value of peralkaline trachytes, particularly NG20, as only the Sr_i isotopes suggest the contamination process undergone by those lavas. Thus, peralkaline felsic lavas, particularly peralkaline trachyte NG20, have been generated through the differentiation process of the associated basaltic lavas, by fractional crystallization, and have been contaminated by the country basement rock, coupled with the effect of Sr (?)enriched fluid, as indicated by their high LREE content (more than 300 times the chondritic normalized values), HFSE (e.g. Zr, Nb, Hf.), and alkali contents (Leat and McDonald, 1984; Bohrson and Reid, 1997; Balcone-Boissard et al., 2009).

Crustal contamination model and peralkalinity of felsic lavas

AFC modelling (after De Paolo, 1981), using partition coefficients from Mahood and Stimac (1990) and a Pan-African gneiss country rock (sample SI: Sr = 112 ppm; $({}^{87}\text{Sr}/{}^{86}\text{Sr})_{9Ma} = 0.749)$ as a contaminant, was also intended. The differentiation of peralkaline trachyte NG18 (0.706072 ± 0.000009) and peralkaline trachyte NG20 (0.710526 ± 0.000008) from the less-evolved lavas requires a Ma/Mc ratio of 0.1 and a fractionation process (F \approx 0.9), involving only feldspar so as to comply with the $({}^{87}\text{Sr}/{}^{86}\text{Sr})_{9Ma}$ values (≈ 0.710) of the considered lavas. Nkouandou et al. (2008) show that major and trace elements of felsic lavas point to their origin as results of fractional crystallization

processes, in spite of the occurrence of a compositional gap, which may be explained by the lack of eruption of lavas of intermediate composition (Bonnefois et al., 1995) or by the high viscosity of the crystal-rich intermediate magmas (Mungall and Martin, 1995). The pronounced Eu anomalies of peralkaline trachytes suggest the strong implication of alkali feldspar in the petrogenetic processes of those lavas, as do the Ba, Sr and Zr contents. The implication of feldspar is also supported by petrographic evidence, as more than 70% of the volume of these lavas consists of sanidine and anorthose. The contamination of lavas by crustal material is difficult to establish using data related to trace elements, as Nb and Ta, which are compatible in crustal rocks, display positive anomalies. Thus, contamination remains selective.

The modal abundance of alkali feldspar (70% of the volume of trachyte lavas) is required to assess the peralkalinity of lavas. As the peralkalinity of lavas is expressed by the ratio (> 1) of (Na₂O + K_2O)/ Al₂O₃, the continued crystallization of alkali feldspars, which fractionate more than 20 % wt of Al₂O₃, remains the best explanation for the peralkaline character of those lavas. This model assumes that increased fractionation of Al₂O₃, coupled with increasing $(Na_2O + K_2O)$ during differentiation, leads to strongly peralkaline residual liquids, as previously proven (Markl et al., 2001; Macdonald and Scaillet, 2006; Ren et al., 2006). Thus, the peralkaline character of Ngaoundéré felsic lavas is linked to the strong fractionation of alkali feldspar and feldspatoid (in peralkaline phonolite), which are highly aluminous and the most abundant phenocryst fractionating phase of all the lavas. However, the lack of an Eu anomaly in the REE patterns (Fig. 4) of peralkaline phonolite NG9, as well as their high Sr (201), Ba (140) and alkali $(Na_2O + K_2O = 16 \text{ wt\%})$ contents, compared to peralkaline trachyte lavas (see Table 4), contradicts the idea of only metaluminous minerals (alkali feldspar. nosean and nepheline) being involved in the lava genesis processes, as those phases might have played only a minor role. These Sr, Ba

and alkali contents of Ngaoundéré felsic lavas may be the evidence of the major role played by hydrothermal fluids (White and Patchett, 1984; McCulloch and Gamble, 1990). As previously noted, the plug and domes occasionally display a strong parallel orientation to the N70E Pan-African direction. Those faults are the possible pathways for alkali metasomatized hydrothermal fluids, which may have contributed significantly to the alkali enrichment of Ngaoundéré felsic lavas. Such hydrothermal event have been noticed in at the Kitongo region, northern Cameroon (Kouske et al., 2012). However, the rapid crystallization of those minerals, and the additional fluid effect, as previously noticed at Mt. Vulture (Italy), where dark rims of minerals from the sodalite group (as nosean in peralkaline phonolite NG9), containing numerous pyrite microcrysts, have been interpreted as the result of a rapid crystallization driven by the exsolution of an S-scavenging fluid phase (Di Muro et al., 2004), may have limited the Eu, Sr and Ba fractionation in this phonolite. Therefore, the peralkalinity of phonolite NG9 is the result of the rapid crystallization of alkali feldspar, sodalite and nepheline, with the additional effect of alkali-enriched fluids. In this respect, the crystallization of Na-clinopyroxene, nepheline and nosean is the result of a continuous alkali enrichment of residual liquid, after more fractionation of the Al₂O₃ component over $(Na_2O + K_2O)$ and volatile fluid (SO₃ up to 10 wt%, Cl up to 0.8 wt% in nosean and F up to 3.5 wt% in apatite). In addition, peralkaline phonolite NG9 is characterized by Sr, Nd and Pb isotopes which exclude any crustal contamination. Among the felsic lavas of the Ngaoundéré series, this lava is particularly depleted in MREE, resulting in a spectacular spoon-shaped pattern (Fig. 4). It contains kaersutite, apatite and particularly abundant titanite phenocrysts (11 % of the volume) which are characterized by high partition distribution coefficients for MREE (Green, 1994; Worner et al., 1983). Slightly depleted Nb and Ta in the spider-like NG9 pattern may be due particularly to titanite crystallization, as this phase is characterized by high Kd values for these elements (Kd_{Nb} = 5, 7 and KD_{Ta} = 15, 5; Green and Pearson, 1987), instead of crustal contamination. Thus, we conclude that the crystallization of those phases is responsible for the slightly depleted Nb and Ta in the multi-element NG9 pattern and its downward MREE patterns. Such spoon-shape patterns also characterize the felsic lavas of the Tchabal Nganha volcano in the Adamawa plateau (Nono et al., 1994; John et al., 2011), northern Cameroon Line (Ngounouno et al., 2003), as well as those on the Fernando de Noronha Island in South Atlantic Ocean (Weaver, 1990).

Conclusions

Field studies, in combination with petrology and geochronology, show that the peralkaline felsic lavas from the Ngaoundéré region represent the final products of the differentiation of the surrounding basaltic lavas through fractional crystallization. Peralkaline trachytes have been contaminated by the basement rock of dioritic composition, coupled with the additional effects of a Sr(?)-enriched fluid phase. Peralkaline phonolite, which is characterized by a low Sr (0.703) isotope, is, however, an exception. The contamination effects remain selective, as it is only evidenced by Sr isotopes. The peralkalinity of felsic Ngaoundéré lavas is clearly due to the pronounced crystallization of aluminous phases (e.g. alkali feldspar in trachyte and alkali feldspar, plus nepheline and nosean, in phonolite), with additional alkali-enriched hydrothermal fluid phases, which have acted through the Pan-African network faults, cutting the basement rocks of the Adamaoua plateau. Thus, the crystallization of enriched Na-phases (aegirine-augite, augite and feldspathoïd) is the result of a continuous enrichment in alkali of residual liquid after the increased fractionation of the Al₂O₃ component over $(Na_2O + K_2O)$, in addition to fluid phase effects. The spoon-shaped patterns of the lavas, particularly phonolite NG9, are due to the crystallization of titanite and amphibole, which preferentially incorporate MREE with Nb and Ta (in titanite).

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References

- Ashwal, L.D., Demaiffe, D., Torsvik, T.H., 2002. Petrogenesis of Neoproterozoic granitoids and related rocks from the Seycheeles: evidence for the case of an Andean-type arc origin. J. Petrology, 43, 45–83.
- Avanzinelli, R., Bindi, L., Menchetti, S., Conticelli, S., 2004. Crystallisation and genesis of peralkaline magmas from Pantelleria Volcano, Italy: an integrated petrological and crystal-chemical study. Lithos, **73**, 41–69.
- Bailey, D.K., 1978. Continental rifting and mantle degassing. In: Neuman, E.R., Ramberg, I.B., (Eds.), (1978). Petrology and geochemistry of continental rifts. Reidel, Dordrecht.
- Bailey, D.K., McDonald, R., 1987. Dry peralkaline felsic liquids and carbon dioxide flux through the Kenya rift zone. In: Mysen, B.O., (Ed.), (2001). Magmatic Processes: Physicochemical Principles. Geochem. Soc. Spec. Pub.
- Balcone-Boissard, H., Baker, D.R., Villemant, B., Boudon, G., 2009. F and Cl diffusion in phonolitic melts: Influence of the Na/K ratio. Chemical Geology, 263, 89–98.
- Bohrson, W.A., Reid, M.R., 1997. Genesis of silicic peralkaline volcanic rocks in an ocean Island setting by crustal melting and open-system processes: Socorro Island, Mexico. Journal of Petrology, 38, 9, 1135–1166.
- Bonnefois, C.C., Provost, A., Albarede, F., 1995. The 'Daly gap' as a magmatic catastrophe. Nature, 378, 270–272.
- Carignan, J., Hild, P., Mevelle, G., Morel, J., Yeghicheyan, D., 2001. Routine analyses of trace elements in geological samples using flow injection and low pressure online liquid chromatography coupled to ICP-MS: a study of geochemical reference materials BR, DR-N, UB-N,AN-G and GH. Geost. News. Lett., 25, 187–198.
- Cousens, B.L., Spera, F.J., Dopson, P.F., 1983. Posteruptive alteration of silicic ignimbrites and lavas, Gran Canaria, Canary Island. Geochimica et Cosmochimica Acta, 57, 631–640.
- DePaolo, D.J., 1981. Trace element and isotopic effects of

combined wall-rock assimilation and fractional cristallization. Earth Planet. Sci. Lett., **53**, 189–202.

- Déruelle, B., Ngounouno, I., Demaiffe, D., 2007. The 'Cameroon Hot Line' (CHL): A unique example of active alkaline intraplate structure in both oceanic and continental lithospheres. C. R. Geoscience, 339, 589–600.
- Di Muro, A., Bonaccorsi, E., Principe, C., 2004. Complex colour and chemical zoning of sodalite-group phases in a hayünophyre lava from Mt. Vulture, Italy. Mineralogical Magazine, 68, 4, 559–582.
- Duggan, M.B., 1988. Zirconium-rich sodic pyroxenes in felsic volcanics from the Warrumbungle Volcano, Central New South Wales, Australia. Mineral. Mag., 52, 491–496.
- Ferguson, A.K., 1977. The natural occurence of aegirineneptunite solid solution. Contrib. Mineral. Petrol., 60, 247–253.
- Ganwa, A.A., Frisch, W., Siebel, W., Ekodeck, G.E., Cosmas, S.K., Ngako, V., 2008. Archean inheritances in the pyroxene-amphibole bearing gneiss of the Méiganga area (Central North Cameroon): Geochemical and 207Pb/206Pb age imprints. C. R. Géoscience, 340, 211–222.
- Gouhier, J., Nougier, J., Nougier, D., 1974. Contribution to the volcanological study of Cameroon (Cameroon Line-Adamawa). Ann. Fac. Sci. Univ. Yaoundé, Cameroon, 17, 3–49. (In French).
- Green, T.H., 1994. Experimental studies of trace-element partitioning applicable to igneous petrogenesis–Sejona 16 years later. Chem. Geol., **117**, 1–36.
- Green, T.H., Pearson, N.J., 1987. An experimental study of Nb and Ta partitioning between Ti-rich minerals and silicate liquids at high pressure and temperature. Geochimica et Cosmochimica Acta, 51, 55–62.
- Halliday, A.N., Davidson, J.P., Holden, P., DeWolf, C., Lee, D.-C., Fitton, J.G., 1990. Trace-element fractionation in plumes and the origine of HIMU mantle beneath the Cameroon line. Nature, **347**, 523–528.
- Halliday, A.N., Dickin, A.P., FallicK, A.E., Fitton, J.G., 1988. Mantle dynamics: a Nd, Sr, Pb, and O isotopic study of the Cameroon line volcanic chain. J. Petrol., 29, 181–211.
- Hamilton, D.L., 1961. Nephelines as crystallization temperature indicators. J. Geol., 22, 433–466.
- John T., Klemd, R., Klemme. S., Jorg, A.P., Hoffmann, E.J., Jun, G., 2011. Nb-Ta fractionation by partial melting at the titanite–rutile transition. Contribution to Mineralogy and Petrology, 161, 35–45.
- Kamgang, P., Njonfang, E., Nono, A., Dedzo, M.G., Tchoua, M.F., 2010. Petrogenesis of a silicic magma system: Geochemical evidence from Bamenda Mountains, NW Cameroon, Cameroon Volcanic Line. Journal of African Earth Sciences, 58, 285–304.
- Kouske, A.P., Suh, C.E., Ghogomu, R.T., Ngako, V., 2012. Na-Metasomatism and Uranium mineralization during a two stage albitization at Kitongo, northern Cameroon: structural and geochemical evidence. International Journal of Geosciences, 3, 258–279.
- Lasserre, M., 1961. Contribution to the geological study of Africa. Study of the eastern part of Adamaoua (Central Cameroon). Bull. Dir. Min. Géol., 4, 1–131. (In French).

- Leat, P.T., Mcdonald, R., 1984. Geochemical evolution of the Menegai Caldeira Volcano, Kenya. Journal of Geophysical Research, 89, 8571–8592.
- Lee, D.-C., Halliday, A.N., Fitton, J.G., Poli, G., 1994. Isotopic variations with distance and time in the volcanic islands of the Cameroon line: evidence for a mantle plume origin. Earth Planet. Sci. Lett., **123**, 119–138.
- Lee, D.-C., Halliday, A.N., Davies, G.R., Essene, E.J., Fitton, J.G., Temdjim, R., 1996. Melt enrichment of shallow depleted mantle: a detailed petrological, trace element and isotopic study of mantle derived xenoliths and megacrysts from the Cameroon line. J. Petrol., 37, 415–441.
- Legendre, C., Maury, R.C., Caroff, M., Guillou, H., Cotten, J., Chauvel, C., Bollinger, C.H. E., Mond, C., Guiller, G., Blais, S., Rossi, P., Savanier, D., 2005. Origin of Exceptionally Abundant Phonolites on Ua Pou Island (Marquesas, French Polynesia): Partial Melting of Basanites Followed by Crustal Contamination. J. Petrology, 46, 9, 1925–1962.
- Macdonald, R., Davies, G.R., Bliss, C.M., Leat, P.T., Bailey, D.K., Smith, R.L., 1987. Geochemistry of high-silica peralkaline rhyolites, Naivasha, Kenya Rift Valley. J. Petrology, 28, 6, 979–1008.
- Macdonald, R., Scaillet, B., 2006. The central Kenya peralkaline province: insights into the evolution of peralkaline salic magmas. Lithos, 91, 59–73.
- Mahood, G.A., Stimac, J.A., 1990. Trace-element partitioning in pantellerites and trachytes. Geochimica et Cosmochimica Acta, 54, 2257–2276.
- Markl, G., Marks, M., Schwinn, G., Sommer, H., 2001. Phase equilibrium constraints on intensive crystallization parameters of the Ilimaussaq Complex, South Greenland. J. Petrology, 42, 2231–2258.
- Marzoli, A., Peccirillo, E.M., Renne, P.R., Bellieni, G., Iacumin, M., Nyobe, J.B., Tongwa, A.T., 2000. The Cameroon Volcanic Line revisited: Petrogenesis of continental basaltic magmas from lithospheric and asthenospheric mantle sources. J. Petrol., 41, 87–109.
- Marzoli, A., Renne, P.R., Piccirillo, E.M., Francesca, C., Bellieni, G., Melfi, A.J., Nyobe, J.B., N'ni, J., 1999.
 Silicic magmas from the continental Cameroon Volcanic Line (Oku, Bambouto and Ngaoundere):
 ⁴⁰Ar-³⁹Ar dates, petrology, Sr-Nd-O isotopes and their petrogenetic significance. Contrib. Mineral. Petrol., 135, 133–150.
- McCulloch, M.T., Gamble, J.A., 1990. Geochemical and geodynamical constraints on subduction zone magmatism. Earth and Planetary Science Letters, **102**, 358–374.
- McDonough, W.F., Sun, S.S., 1995. The composition of the Earth. Chemical Geology, 120, 223–253.
- Moreau, C., Regnoult, J.-M., Déruelle, B., Robineau, B., 1987. A new tectonic model for the Cameroon Line, central Africa. Tectonophysics, **139**, 317–334.
- Morimoto, N., Fabries, J., Ferguson, A.K., Ginzburg, I.V., Ross, M., Seifert, F.A., Zussman, J., Aoki, K., Gottardi, G., 1988. Nomenclature of pyroxenes, IMA. Mineralogical Magazine, 52, 533–550.
- Mungall, J.E., Martin, R.F., 1995. Petrogenesis of basaltcomendite and basalt-pantellerite suites, Terzeira, Azores, and some implications for tha origin of the ocean-Islands rhyolites. Contribution to Mineralogy and Petrology, **119**, 43–55.

- Ngounouno, I., Déruelle, B., Demaiffe, D., Montigny, R., 2003. Petrology of the Cenozoic volcanism in the Upper Benue valley, northern Cameroon (Central Africa). Contrib. Mineral. Petrol., **145**, 87–106.
- Nielsen, T.F.D., 1979. The occurrence and formation of Tiaegirines in peralkaline syenites. An example from the Tertiary ultramafic alkaline Gardiner complex, East Greenland. Contrib. Mineral. Petrol., 69, 235–244.
- Nkouandou, O.F., Ngounouno I., Déruelle, B., 2010. Geochemistry of recent basaltic lavas from the north and east of Ngaoundéré zones (Cameroon, Adamawa Plateau, Central Africa): petrogenesis and the nature of the source. Int. J. Biol. Chem. Sci., 4, 4, 984–1003. (In French).
- Nkouandou, O.F., Ngounouno, I., Déruelle, B., Ohnenstetter, D., Montigny, R., Demaiffe, D., 2008. Petrology of the Mio-Pliocene Volcanism to the North and East of Ngaoundéré (Adamawa-Cameroon). C. R. Géoscience, 340, 27–38.
- Nkouandou, O.F., Temjim, R., 2011. Petrology of spinel lherzolite xenoliths and host basaltic lava from Ngao Voglar volcano, Adamawa Massif (Cameroon Volcanic Line, West Africa): equilibrium conditions and mantle characteristics. Journal of Geosciences, 56, 375–387.
- Nono, A., Déruelle, B., Demaiffe, D., Kambou, R., 1994. Tchabal Nganha volcano in Adamawa (Cameroon): petrology of a continental alkaline lava series. J. Volcanol. Geotherm. Res., 60, 147–178.
- Pouchou, J.L., Pichoir, F., 1991. Quantitative analysis of homogeneous or stratified microvolumes applying the model "PAP". In: Heinriche, D.E. (Eds.), (1991). Electron Probe Quantification. Plenum Press.
- Ren, M., Omenda, P.A., Anthony, E.Y., White, J.C., Macdonald, R., Bailey, D.K., 2006. Application of the QUILF thermobarometer to the peralkaline trachytes and pantellerites of the Eburru volcanic complex, East African Rift, Kenya. Lithos, **91**, 109–124.
- Taylor, R.P., Strong, D.F., Kean B.F., 1980. The Topsails igneous complex: Silurian-Devonian peralkaline magmatism in western Newfoundland. Can. J. Earth. Sci., 17, 425–439.
- Tchameni, R., Pouclet, A., Penaye, J., Ganwa, A.A., Toteu, S.F., 2006. Petrography and geochemistry of the Ngaoundéré Pan-African granitoïds in central north Cameroon: implications for their sources and geological setting. Journal of African Earth Sciences, 44, 511–529.
- Temdjim, R., Njilah, I.K., Kamgang, P., Nkoumbou, C., 2004. New data on the felsic lavas from Ngaoundéré (Adamawa, Cameroon Line): K–Ar chronology and petrology. Afr. J. Sci. Technol., 5, 113–123. (In French).
- Thompson, G.M., Smith, I.E.M., Malpas, J.G., 2001. Origin of oceanic phonolites by crystal fractionation and the problem of the Daly gap: an example from Rarotonga. Contributions to Mineralogy and Petrology, 142, 336–346.
- Thornton, C.P., Tuttle, O.F., 1960. Chemistry of igneous rocks. I. Differentiation Index, American Journal of Science, 258, 664–684.
- Wandji, P., Tchokona, D.S., Bardintzeff, J.-M., Bellon, H., Platevoet, B., 2008. Rhyolites of the Mbépit Massif in

the Cameroon Volcanic Line: an early extrusive volcanic episode of Eocene age. Miner Petrol., **94**, 271–286.

- Weaver, B.L., 1990. Geochemistry of highly-undersaturated ocean island basalt suites from the south Atlantic Ocean: Fernando de Noronha and Trindade islands. Contrib. Mineral. Petrol., 105, 502–515.
- Wellman, T.R., 1970. The stability of sodalite in a synthetic syenite plus aqueous chloride fluid system. J. Petrol., 11, 149–171.
- White, W.M., Patchett, J., 1984. Hf-Nd-Sr isotopes and incompatible element abundances in island arcs: implications for magma origins and crust mantle evolution. Earth and Planetary Science Letters, 67, 167–185.
- Woolley, A.R., Platt, R.G., 1988. The peralkaline nepheline syenites of the Junguni intrusion, Chilwa province, Malawi. Mineralogical Magazine, 52, 425–433.

- Wörner, G., Beusen, J.-M., Duchateau, N., Gijbels, R., Schmincke, H.-U., 1983. Trace element abundances and mineral/melt distribution coefficients in phonolites from the Laacher See Volcano (Germany). Contrib. Mineral. Petrol., 84, 152–173.
- Yokoyama, T., Aka, T.F., Kusakabe, M., Nakamura, E., 2007. Plume-lithosphere interaction beneath Mt. Cameroon volcano, West Africa: Constraints from ²³⁸U-²³⁰Th-²²⁶Ra and Sr-Nd-Pb isotope systematic. Geochimica et Cosmochimica Acta, **71**, 1835–1854.

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